

## Synthesis Of 2-Benzamidomethyl-5-Substituted Amino-1,3,4-Thiadiazoles-2,5-Disubstituted 1,3,4- Oxadiazole and 4,5-Disubstituted 1,2,4-Triazole-3- Thiol

**K. M. Daoud      A. N. Ali      A. A. Ahmed**

Department of Chemistry / College of Education  
University of Mosul

**Received**  
**08 / 05 / 2011**

**Accepted**  
**01 / 06 / 2011**

- , ,      - , ,  
)  
( , ) -      ( )  
( , )      ( , )  
( , )      .( , ) -  
.  
( , )      -4,3,1 ( , )      - , ,  
-4,2,1      .( )      -4,3,1      ( )  
( )      -      ( )  
( )      ( )  
( )      - -      -4,2,1      -  
.

### Abstract

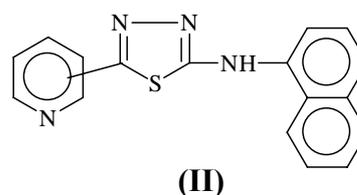
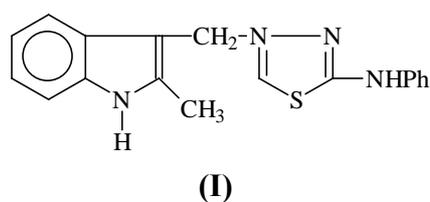
In this paper the synthesis of some substituted 1,3,4-oxadiazoles , 1,3,4-thiadiazoles and 1,2,4-triazoles starting from amino acid and benzoyl chloride is reported. Treatment of glycine or alanine with benzoyl chloride in presence of sodium carbonate gave keto-acid(1,2), which were converted to oxazolinone (3,4) by their reaction with acetic

anhydride. Oxazolinone (3,4) were treated with hydrazine hydrate to give acid hydrazides (5,6). Acid hydrazides were converted to substituted thiosemi- carbazides (7,8) by their reaction with phenyl isothiocyanate. Substituted thiosemicarbazides (7,8) were treated with sodium hydroxide and concentrated sulphuric acid give substituted 1,2,4-triazoles (9,10), 1,3,4-thiadiazoles (11,12) respectively while treatment of (7) with mercuric oxide gave 1,3,4-oxadiazoles (13). Treatment of triazoles (9,10) with 4-hydroxybenzaldehyde gave substituted triazoles (14,15). 1-Substituted thiosemicarbazide (16) was obtained from acid hydrazide (5), cyclized with sodium hydroxide to 5-Substituted-1,2,4-triazole-3-thiol (17). The structures of the synthesized compounds were confirmed by physical and spectral means.

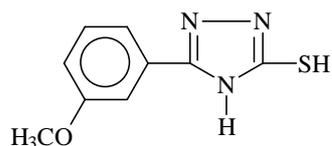
### Introduction

Five membered ring heterocyclic 1,3,4-oxadiazoles, 1,3,4-thiadiazole and 1,2,4-triazoles and their derivatives are considered as an important class of compounds because of their diversified biological applications<sup>(1)</sup>. 1,3,4-oxadiazole derivatives showed antibacterial<sup>(2,3)</sup>, anti-inflammatory<sup>(4)</sup>, antitubercular<sup>(5)</sup>, anticonvulsant<sup>(6)</sup>, antimalarial<sup>(7)</sup> and antifungal agents<sup>(8)</sup>.

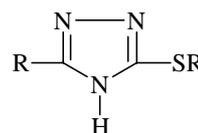
1,3,4-thiadiazole exhibit various biological activities as anticancer<sup>(9)</sup>, anticonvulsant<sup>(10)</sup>, antibacterial<sup>(11)</sup>, anti-inflammatory<sup>(12)</sup> and antifungicidal<sup>(13)</sup>. 1,2,4-triazole derivatives showed biological effects, such as antifungal<sup>(14)</sup>, anticancer<sup>(15)</sup>, antibacterial<sup>(16)</sup>, anticonvulsant<sup>(17)</sup> and anti-inflammatory<sup>(18)</sup>. The synthesis of substituted 1,3,4-oxadiazoles was achieved by condensation of acyl hydrazine with acetic acid in phosphorous oxychloride to give 2,5-disubstituted 1,3,4-oxadiazole<sup>(19)</sup>. 2-Substituted 1,3,4-oxadiazole were synthesized from acid hydrazides by their reaction with carbon disulfide in ethanolic potassium hydroxide<sup>(20,21)</sup>. 1,3,4-oxadiazoles were synthesized by cyclization of substituted thiosemicarbazide by phosphoric acid<sup>(22)</sup>, concentrated sulphuric acid<sup>(23)</sup> or methyl sulfonate<sup>(24)</sup>. 1,3,4-Thiadiazoles (I) and (II) were synthesized from substituted thiosemicarbazide by phosphoric acid<sup>(22)</sup> and concentrated sulphuric acid<sup>(25)</sup> respectively.



1,2,4-Triazole derivatives were synthesized from thiosemicarbazides by their reaction with sodium hydroxide solution as compound (III)<sup>(26)</sup>. 5-alkyl-1,2,4-triazole-3-thiol was treated with alkyl halides to give alkylthio derivative (IV)<sup>(27)</sup>.



(III)



(IV)

## Experimental

### N-Benzoyl amino acid (1,2)<sup>(28)</sup>

A mixture of amino acid (glycine or alanine) (0.12 mole), benzoyl chloride (22.4 g, 0.16 mole) and sodium carbonate (20 g) in water (200 ml) was refluxed for 6 hours, the mixture was cooled to room temperature and acidified with concentrated hydrochloric acid (pH=6) and left to stand over night. The precipitate was filtered off, washed with cold water and recrystallized from water. Compound (1), X=H, m.p. 111-112 °C; yield 80%, white crystals. IR, KBr  $\nu_{\text{cm}^{-1}}$ , 1717(C=O), 2957(C-H aliphatic), 1070(C-O), 3418(N-H). Compound (2), X=CH<sub>3</sub>; m.p. 116 °C; yield 85%; white crystals. IR, KBr  $\nu_{\text{cm}^{-1}}$ , 1100(C-O), 1686(C=O), 3071(C-H aromatic), 3449(N-H), 2900(C-H aliphatic).

### 2-phenyl-4-methyl/H- $\Delta^2$ -5-oxazolinone(3,4)<sup>(29)</sup>

Compound (1,2) (0.018 mole) was dissolved in acetic anhydride (15ml). The mixture was heated at 70 °C for 1 hour, after cooling ether-petroleum ether (10:50 ml) were added then the mixture stirred for 30 minutes. Compound (3) was isolated as oil, compound (4) as white solid was recrystallized from petroleum ether-ether (1:1). Compound (3); X=H, yellow oil, yield 55%, IR, KBr  $\nu_{\text{cm}^{-1}}$ , 1755(C=O), 1644(C=N), 2942(C-H, aliphatic), 3025(C-H, aromatic), 1046, 1126(C-O). Compound (4); X=CH<sub>3</sub>, m.p. 98 °C, yield 75%, IR, KBr  $\nu_{\text{cm}^{-1}}$ , 1746(C=O), 1618(C=N), 3005(C-H, aromatic), 1050(C-O).

### Acid Hydrazide (5,6)<sup>(29)</sup>

A mixture of compound (3or4)(0.035mole) in dioxane (10ml) and hydrazine hydrate (1.12 g, 0.035mole) was heated for 1 hour. The solvent was evaporated to give compound (5) as an oil whereas compound(6) was dissolved in ethanol and crystallized by water, filtered and dried. Compound (5); X=H, yield 62%, yellow oil.; IR, KBr  $\nu_{\text{cm}^{-1}}$ , 1650(C=O), 3453(N-H). Compound (6); X=CH<sub>3</sub>, m.p. 260d., yield 75%, brown, IR, KBr  $\nu_{\text{cm}^{-1}}$ , 1680(C=O), 3056(C-H, aromatic), 3311(N-H).

### Substituted thiosemicarbazide (7,8)<sup>(29)</sup>

To acid hydrazides (5or6) (0.0025mole) in methanol(10ml), phenyl isothiocyanate (0.0025mole) in methanol (10ml) was added. The mixture

then refluxed for 2 hours. The solvent was evaporated under reduced pressure, compound (7) was recrystallized from ethanol. Compound (7); Ar=Ph, R=PhCO-NH, m.p.182-184 °C, yield 72%, pale green crystals; IR,KBr  $\nu\text{cm}^{-1}$ , 3444(N-H), 3009(C-H,aromatic), 2988(C-H,aliphatic), 1190(C=S),1645(C=O). Compound (8); Ar=4-OCH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>; R=PhCONH, yield 44%, green oil.; IR,KBr  $\nu\text{cm}^{-1}$ , 1663(C=O), 1115(C=S),3418(N-H).

### 5-Substituted-4-aryl-1,2,4-triazole-3-thiol(9,10)<sup>(29)</sup>

A mixture of substituted thiosemicarbazide (7or8) (0.0014mole) in sodium hydroxide solution 1N (10ml) was heated at 80 °C for 1 hour, water (15 ml) then was added then acidified with dilute hydrochloric acid (pH=4.5). The precipitate was filtered off and recrystallized from ethanol. Compound (9); Ar=Ph ;R=RCONH, m.p. 195-197 °C, yield 75%, pale brown crystals.; IR, KBr  $\nu\text{cm}^{-1}$ , 1630(C=N), 3452(N-H), 2940(C-H, aliphatic), 1122(C=S). Compound (10); Ar=4-OCH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>; R=RCONH, m.p.154 °C yield 81%, pale green crystals.; IR,KBr  $\nu\text{cm}^{-1}$ , 3451(N-H), 2910 (C-H,aliphatic), 1218(C=S), 1623(C=N).

### 2-Substituted-5-arylamino-1,2,4-triazole-3-thiol(11,12)<sup>(29)</sup>

Substituted thiosemicarbazide (7or8) was dissolved in concentrated sulphuric acid (1ml) then stirred at room temperature for 1 hour, cool water (25ml) was added, the precipitate was filtered,dried and recrystallized from ethanol. Compound (11); Ar=Ph; R=PhCONH, m.p.140-142 °C yield 71%, pale brown crystals.; IR,KBr  $\nu\text{cm}^{-1}$ ,3417 (N-H), 2983(C-H,aliphatic), 3122(C-H,aromatic), 1633(C=O), 1602(C=N), U.V.  $\lambda_{\text{max}}$  225nm. Compound (12); Ar=4-OCH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>; R=PhCONH, m.p. 222-224 °C yield 56%, pale brown.; IR, KBr  $\nu\text{cm}^{-1}$ , 3443(N-H), 1629 (C=O),1570(C=N).

### 2-Substituted-5-phenylamino-1,3,4-oxadiazole(13)<sup>(30)</sup>

A mixture of substituted thiosemicarbazide (7)(0.20 g,0.001mole) in methanol(25ml) and mercuric oxide (0.24g,0.001 mole) was refluxed for four hours, the mixture was filtered while hot, the solvent then evaporated under reduced pressure to give solid product, recrystallised from ethanol., m.p.135-137 °C d., yield 51%, brown crystals, IR, KBr  $\nu\text{cm}^{-1}$ ,3393(N-H), 3050(C-H,aromatic), 2852(C-H,aliphatic),1620(C=N),1088(C-O).

### 3,5-Disubstituted-1,2,4-triazole(14,15)<sup>(31)</sup>

1,2,4-triazole (9,10) (0.002mole) was dissolved in ethanol (20ml), the solution cooled in ice-bath, 4-carboxybenzaldehyde (0.002mole) was added, the mixture then kepted in ice-bath for 4-6 hours, the precipitate was filtered off, dried and recrystallized from ethanol-water. Compound (14); Ar=Ph; R=PhCONH, m.p.187-188 °C yield 81%, brown crystals.; IR,KBr  $\nu\text{cm}^{-1}$ ,3305(O-H), 1294(C=S), 1612(C=N). Compound (15); Ar=4-OHC<sub>6</sub>H<sub>4</sub>; R=PhCONH, m.p.106 °C yield 61%, yellowish crystals.;

IR, KBr  $\nu_{\text{cm}^{-1}}$ , 1599(C=N), 1156(C=S), 3300(OH), 2956 (C-H, aliphatic), 3165(C-H, aromatic).

### **Substituted thiosemicarbazide (16)<sup>(32)</sup>**

A mixture of hydrazide (5) (1.48 g, 0.01 mole), ammonium thiocyanate (2.28 g, 0.03 mole) and concentrated hydrochloric acid (4 ml) in ethanol (50 ml) was refluxed for 22 hours. The solvent was evaporated under reduced pressure to give oily product. Yield 65%, colour pale violet IR, KBr  $\nu_{\text{cm}^{-1}}$ , 3444(N-H), 2950(C-H, aliphatic), 1664(C=O), 1044(C=S).

### **5-Substituted-1,2,4-triazole-3-thiol(17)<sup>(33)</sup>**

A mixture of substituted thiosemicarbazide (16) (0.412 g, 0.002 mole) in 4% sodium hydroxide solution (25 ml) was refluxed for three hours, the mixture was cooled acidified with 10% hydrochloric acid. The precipitate then filtered and recrystallized from ethanol-water., m.p. 200 °C d., yield 62%, pale green crystals, IR, KBr  $\nu_{\text{cm}^{-1}}$ , 3419(N-H), 1635(C=N), 3090 (C-H, aromatic), 2980(C-H, aliphatic), 1125(C=S).

### **3,5-Disubstituted-4-amino-1,2,4-triazole(18)<sup>(34)</sup>**

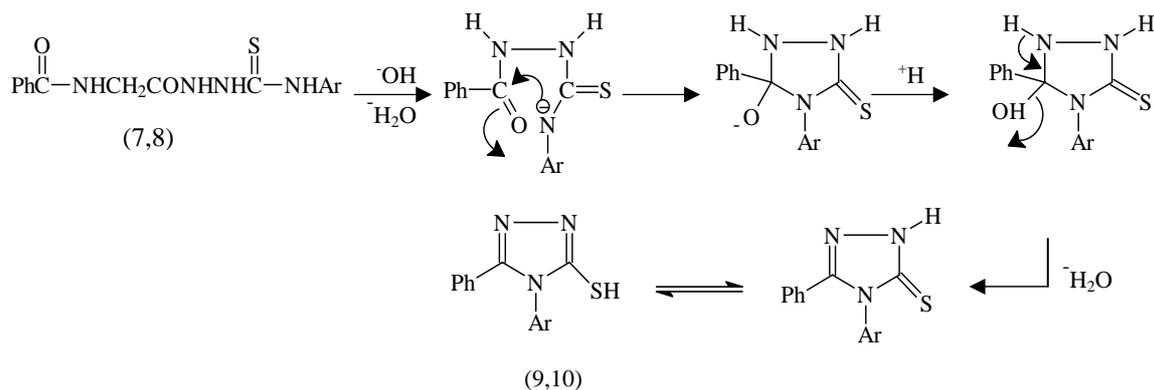
Acid hydrazide (6) (0.32 g, 0.002 mole) was heated at 130-150 °C for 1 hour, water (50 ml) was added then refluxed for 15 minutes. A solid product was formed on cooling. m.p. 240 °C d. pale brown crystals, yield 70% IR, KBr  $\nu_{\text{cm}^{-1}}$ , 3318(N-H), 3058(C-H, aromatic), 2927(C-H, aliphatic), 1596 (C=N), 1650(C=O).

## **Result and discussion**

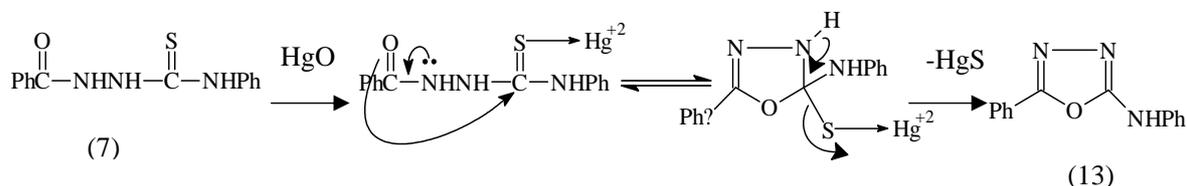
The synthesis of some substituted 1,3,4-oxadiazole, 1,3,4-thiadiazoles and 1,2,4-triazoles is reported scheme-1-. Amino acid (glycine and alanine) was treated with benzoyl chloride to give N-benzoyl amino acids (1,2). IR spectra of compound (1) KBr  $\text{cm}^{-1}$  3418(N-H), 2957 (C-H, aliphatic), 1717(C=O) and 1070(C-O), whereas compound (2) IR spectra showed absorption bands at  $\text{cm}^{-1}$  3449 (N-H), 3071 (C-H, aromatic), 2900(C-H, aliphatic), 1686(C=O) and 1100(C-O). U.V. spectra  $\lambda_{\text{max}}$  234, 205 nm. Compound (1,2) were treated with acetic anhydride to give oxazolinones (3,4). The IR spectra of compounds (3,4) showed absorption bands within the range 3050-3025  $\text{cm}^{-1}$  that were attributed to (C-H, aromatic), 1746-1755  $\text{cm}^{-1}$  (C=O), 1634  $\text{cm}^{-1}$  (C=N) 1050-1126  $\text{cm}^{-1}$ . U.V. spectra  $\lambda_{\text{max}}$  217, 242 nm. Compounds 3,4 were converted into the acid hydrazides (5,6) by their reaction with hydrazine hydrate in dioxane. The IR spectra for compounds (5,6) showed an absorption band between 3311-3453  $\text{cm}^{-1}$  (N-H), 1650-1680  $\text{cm}^{-1}$  (C=O), U.V. spectra  $\lambda_{\text{max}}$  216, 206 nm. Acid hydrazide (5,6) were reacting with phenyl isothiocyanate in methanol to give substituted thiosemicarbazide (7,8), IR spectra of compounds (7,8) showed absorption band at 3418-3444  $\text{cm}^{-1}$  due to (N-H), 1645-1663  $\text{cm}^{-1}$  (C=O), 1115-1190 (C=S), U.V. spectra  $\lambda_{\text{max}}$  (242, 221 nm). Substituted thiosemicarbazides (7,8) were treated with sodium hydroxide solution,

concentrated sulphuric acid to give 1,2,4-triazoles (9,10) and 1,3,4-thiadiazole (11,12) respectively. whereas treatment of compound (7) with mercuric oxide give 1,3,4-oxadiazole(13). The IR spectra of compounds (9,10) showed absorption band at  $3451-3452\text{cm}^{-1}$  due to (N-H),  $1623-1630\text{cm}^{-1}$  (C=N),  $1122-1218$  (C=S). U.V. spectra  $\lambda_{\text{max}}$  (285,212 nm). IR spectra of compounds (11,12) the absorption band of (N-H) appeared at  $3417-3443\text{cm}^{-1}$  (C=O) at  $1629-1633\text{cm}^{-1}$ , (C=N) at  $1570-1602\text{cm}^{-1}$ . U.V. spectra  $\lambda_{\text{max}}$  (225,200 nm).

The mechanism for the conversion of (7,8) to (9,10) as follows<sup>(35)</sup>:



The IR spectra of compound(13) showed absorption bands at  $3393\text{cm}^{-1}$  (N-H),  $3050\text{cm}^{-1}$  (C-H, aromatic),  $1620$ (C=N),  $1088$ (C-O). U.V. spectra  $\lambda_{\text{max}}$  325nm. The proposed mechanism for the conversion of (7) to (13) as follows:



The reaction of triazoles (9,10) with 4-hydroxybenzaldehyde gave substituted 1,2,4-triazoles(14,15). The IR spectra of of compounds (14,15) showed absorption bands at  $3305-3300$  (O-H),  $1599-1612$ (C=N),  $1156-1294$  (C=S). U.V. spectra  $\lambda_{\text{max}}$ . 202,257nm. The reaction of acid hydrazide (5) with ammonium thiocyanate and hydrochloric acid in ethanol afforded substituted thiosemicarbazide (16) which cyclized to 1,2,4-triazole (17) by aqueous sodium hydroxide. The IR spectra of compound (16) showed absorption bands at  $3444\text{cm}^{-1}$  (N-H),  $1664\text{cm}^{-1}$  (C=O),  $1044\text{cm}^{-1}$  (C=S). U.V. spectra  $\lambda_{\text{max}}$ . 204nm. Whereas compound (17) the absorption band for (N-H) appeared at  $3419\text{cm}^{-1}$ , for (C=N) at  $1635\text{cm}^{-1}$  and for (C=S)  $1125\text{cm}^{-1}$ . U.V. spectra  $\lambda_{\text{max}}$ . 212nm. Acid hydrazide (6) was converted into triazole (18) when heated at  $130-150\text{ }^\circ\text{C}$  for one hour. The IR spectra of compound (18) showed absorption band at  $3318\text{cm}^{-1}$  (N-H),  $1630\text{cm}^{-1}$ (C=O),  $1596\text{cm}^{-1}$  (C=N). U.V. spectra  $\lambda_{\text{max}}$ . 203nm.



## References

- 1) S.Wagle, A.A.Vasudeva and N.K.Suchetha, 2008, Indian J. Chem., 47B,439
- 2) H.Xin-Ping, Z.Lin-Mei and Z.Zi-yi, 1999, Indian J. Chem., 38B, 1066.
- 3) J.Salimon, N.Salih, A.Hameed, H.Ibraheem, E.Yousif, 2010, J. of Applied Science Research 6(7), 866.
- 4) S.Sharma, V.S.Kishor and A.Kumar, 2002, Indian J. Chem., 41B, 2647.
- 5) S.Dhoel, A.S.Bhimani, R.C.Khunt and A.R.Parikh, 2005, Indian J. Heterocyclic Chemistry,15,63.
- 6) A.Omar, M.E.Mohsen, O.M.Aboulwafa, 1984, J.Heterocycl. Chem., 21,1415.
- 7) M.P.Hutt, E.F.Werbet, L.M.,1970, J. Heterocycl. Chem.,7,511.
- 8) A.Rauf, S.Sharma, S.Gangal, 2008, Chin.Chem.Let.
- 9) J.Matysiak, 2007, Eur.J.Med. Chem., 42, 940.
- 10) K.Srivastava and S.N.Pandeya,1993, Bioorg and Med.Chem., 3, 547.
- 11) A.Foroumadi, S.Mansouri, Z.Kirani and A.Rahman, 2003, Eur.J. Med. Chem., 38,851.
- 12) M.Amir and K.Shikha, 2004, Eur.J.Med. Chem., 39,535.
- 13) R.K.Khare, H.Sing and A.K.Srivastava, 2007, Ind.J.Chem.46B, 875.
- 14) X.Collin, A.Sauleau and J.Goulon, 2003, Bioorg. Med. Chem. Lett.,13, 2601.
- 15) B.S.Holla and Poojary, 2003, Eur.Med. Chem., 38,759.
- 16) G.Sun, X.P.Hui, P.F.Xu, Z.Y.Zhang and Z.W.Guan, 2007, J. Chinese Chem., Soc., 54,795.
- 17) A.Cansiz, M.Koparir and A.Demirdag, 2004, Molecules, 9,204.
- 18) A.Alkan, H.Yukseh, O.Gursoy-Kol and M.Calapoglu, 2008, Molecules,13,107.
- 19) E.H.El-Tamaty, M.E.Abdel-Fattah and I.M.El-Deen, 1996, Indian Journal of Chemistry, Vol.35B,1067.
- 20) O.S.Moustafa, M.Z.A.Badr and T.I-El-Emary,2002,Bull.Korean Chem. Soc., Vol., 23, No. 4,568.
- 21) A.O.Maslat, M.Abussad, H.Tashtoush and M.Al-Talib, 2002, Pol.J. Pharmacol.,Vol., 54,55.

- 22) A.Varvaresou, P.T.Siatra, T.A.Dalla and K.Tsantili, 1998, *IL Farmaco.*, 53,320.
- 23) K.Zamami, K.Faghihi and M.S.Mehranijani, 2003, *Pol.J. Pharmacol*, 55,1111.
- 24) E.Palaska, G.Sahin., P.Kelicent, N.T.Durlu and E.Altinok, 2002, *IL Farmaco.*, 57,101.
- 25) K.Zamout, K.Faghihi, T.Tofighi and M.R.Shariatzadeh, 2004, *Turk. J. Chem.*, 28,95.
- 26) L.Labonauskas, E.Udrenaite, P.Gaidelic and A.Brukstus, 2004, *IL Farmaco.*,59.255.
- 27) R.B.Pathak, B.Jahan and S.C.Bahel, 1980, *Bokin*, Vol.8,58.
- 28) A.I.Vogel, (1965), "Atext-book of practical organic chemistry"3rd. Edn., Longmans green and Co.Ltd, London, p.980.
- 29) O.Pintilie, L.Profirc, V.Sunel, M.Popa and A.Pui, 2007, *Molecules*, 12, 103-113.
- 30) R.S.Sharma and S.C.Bahel,(1982), *J.Indian Chem.Soc.*,LIX,877.
- 31) R.B.Pathak, V.Srivatav and S.C.Bahel, (1982), *J. Indian Chem. Soc.*, LIX,776.
- 32) M.T.Wu, (1972), *J. Heterocyclic Chem.*, 9,31.
- 33) S.M.El-Khawass and N.S.Habib, (1989), *J. Heterocyclic Chem.*, 26, 177.
- 34) A.O.Mohammad, (2001), "Synthesis of some new coumarin compounds and the study of their biological activity", M.Sc. Thesis, University of Mosul-Iraq.
- 35) J.P.Henchart, R.Houssin and B.Lablanche, 1977, *J. Heterocyclic. Chem.*,14,615.